

LETTERS  
TO THE EDITOR

## Reaction of 2-Phenyl-1,3,2-benzodioxaphosphinan-4-one with Diethyl Benzyldenemalonate

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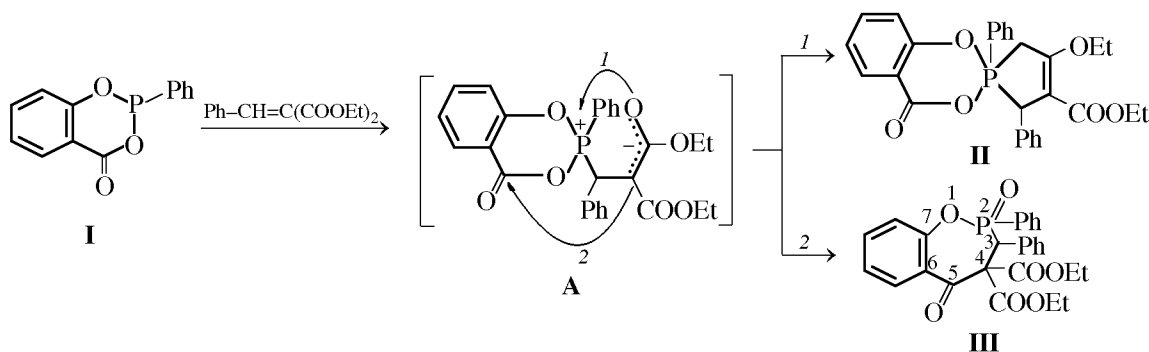
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Reactions of 2-substituted 1,3,2-benzodioxaphosphinan-4-ones **I** (salicyl phosphites) with carbonyl compounds activated by electron-acceptor substituents, as well as with Schiff bases, provide a convenient route to seven-membered phosphorus-containing heterocycles: 1,3,2- and 1,4,2-dioxa- or 1,3,2- and 1,4,2-oxazaphosphepines [1–3]. In these reactions compounds **I** can act as both nucleophile and electrophile, and the process is characterized by high regio- and stereoselectivity.

Phosphorus(III) compounds are known to readily react with ylidene derivatives of  $\beta$ -dicarbonyl compounds having activated C=C bonds, yielding the corresponding phosphoranes [4]. We are the first to report that ylidene derivatives of diethyl malonate are capable of being inserted into the dioxaphosphinane ring of **I**. Diethyl benzyldenemalonate readily reacts with phenylphosphonite **I** along two pathways, yielding five- and four-coordinate phosphorus deriva-

tives **II** ( $\delta_P$  –22.6 ppm) and **III** ( $\delta_P$  38.4 and 38.1 ppm), respectively. Heating of the reaction mixture for a short time leads to disappearance of the phosphorane signal. In keeping with the  $\delta_P$  value, compound **II** is likely to contain two P–C bonds and one five-membered ring. Oxaphosphepine **III** is formed with high stereoselectivity: the intensity ratio of the signals with  $\delta_P$  38.4 and 38.1 ppm is 3:1. Its structure was unambiguously determined by  $^1H$  and  $^{13}C$  NMR spectroscopy.

Presumably, the reaction scheme involves nucleophilic attack by the phosphorus on the  $\beta$ -carbon atom of diethyl benzyldenemalonate to produce dipolar ion **A**. The latter is stabilized either via attack by the oxygen atom on the phosphorus with reversible formation of phosphorane **II** (pathway 1) or through nucleophilic substitution at the carbonyl carbon atom (pathway 2) to give a new phosphorus-containing ring, 1,2-benzoxaphosphepine **III**.



This reaction is the first example of participation of compounds with a double C=C bond in dioxaphosphinane ring expansion.

**Diethyl 2,5-dioxo-2,3-diphenyl-1,2λ<sup>5</sup>-benzoxaphosphepine-4,4-dicarboxylate (III)** was obtained by heating a mixture of 5 g of phosphonite **I** and 5.02 g of diethyl benzylidenemalonate in 10 ml of methylene chloride for 2 h at 40°C. The product was isolated by precipitation (the reaction mixture was poured into ether). Yield 70%, mp 166°C. IR spectrum,  $\nu$ , cm<sup>-1</sup>: 1690 (C=O), 1740 and 1770 (COO), 1290 (P=O). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm (*J*, Hz): 4.47 d (3-H, <sup>2</sup>*J*<sub>P,CH</sub> = 27.5), 4.29 m and 4.14 m (OCH<sub>A</sub>H<sub>B</sub>, ABX<sub>3</sub> system, <sup>3</sup>*J*<sub>BX</sub> = 7.0, <sup>3</sup>*J*<sub>AX</sub> = 7.0, <sup>3</sup>*J*<sub>AB</sub> = 10.8), 3.79 m and 3.84 m (OCH<sub>A</sub>H<sub>B</sub>, ABX<sub>3</sub> system, <sup>3</sup>*J*<sub>AX</sub> = 7.1, <sup>3</sup>*J*<sub>BX</sub> = 7.1, <sup>3</sup>*J*<sub>AB</sub> = 10.7), 1.50 t and 0.74 t (CH<sub>3</sub>, X<sub>3</sub> part of the ABX<sub>3</sub> system, <sup>3</sup>*J*<sub>A(B)X</sub> = 7.0, <sup>3</sup>*J*<sub>(A)BX</sub> = 7.1). <sup>13</sup>C NMR spectrum,  $\delta$ <sub>C</sub>, ppm (*J*, Hz): 50.84 d.d.t (C<sup>3</sup>, <sup>1</sup>*J*<sub>CP</sub> = 84.2, <sup>1</sup>*J*<sub>CH</sub> = 126.9, <sup>3</sup>*J*<sub>CH</sub> = 4.4), 74.58 d.d (C<sup>4</sup>, <sup>2</sup>*J*<sub>CP</sub> = 1.8, <sup>2</sup>*J*<sub>CH</sub> = 5.4), 188.05 d.d.d (C<sup>5</sup>, <sup>3</sup>*J*<sub>CP</sub> = 0.96, <sup>3</sup>*J*<sub>CH</sub> = 5.8–5.9, <sup>3</sup>*J*<sub>CH</sub> = 4.6–4.8), 126.41 m (C<sup>6</sup>, <sup>3</sup>*J*<sub>CP</sub> = 1.54), 148.74 d.d.d.d (C<sup>7</sup>, <sup>2</sup>*J*<sub>CP</sub> = 8.1, <sup>3</sup>*J*<sub>CH</sub> = 9.7–9.8, <sup>2</sup>*J*<sub>CH</sub> = 4.0, <sup>4</sup>*J*<sub>CH</sub> = 1.76), 165.80 d.t.d (COO, <sup>3</sup>*J*<sub>CP</sub> = 14.1, <sup>3</sup>*J*<sub>CH</sub> = 3.5, <sup>3</sup>*J*<sub>CH</sub> = 2.5), 165.79 m (COO, <sup>3</sup>*J*<sub>CP</sub> = 6.6, <sup>3</sup>*J*<sub>CH</sub> = 3.5), 62.71 t.q and 62.17 t.q (OCH<sub>2</sub>, <sup>1</sup>*J*<sub>CH</sub> = 148.8, <sup>3</sup>*J*<sub>CH</sub> = 4.5), 13.59 q.t and 13.06 q.t (CH<sub>3</sub>, <sup>1</sup>*J*<sub>CH</sub> = 127.5,

<sup>3</sup>*J*<sub>CH</sub> = 2.6). Found, %: C 65.48; H 4.94; P 6.10. C<sub>27</sub>H<sub>25</sub>O<sub>7</sub>P. Calculated, %: C 65.85; H 5.08; P 6.30.

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